Acta Crystallographica Section E
Structure Reports Online

ISSN 1600-5368

## Kanji Kubo* and Akira Mori

Institute of Advanced Material Study, 86, Kyushu University, Kasuga-koen, Kasuga, Fukuoka 816-8580, Japan

Correspondence e-mail:
kubo-k@cm.kyushu-u.ac.jp

## Key indicators

Single-crystal X-ray study
$T=296 \mathrm{~K}$
Mean $\sigma(\mathrm{C}-\mathrm{C})=0.005 \AA$
$R$ factor $=0.059$
$w R$ factor $=0.193$
Data-to-parameter ratio $=21.1$

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.
(C) 2001 International Union of Crystallography Printed in Great Britain - all rights reserved

## 4-Cyanophenyl 4-n-dodecanyloxybenzoate

The title compound, $\mathrm{C}_{26} \mathrm{H}_{33} \mathrm{NO}_{3}$, has a phase sequence of crystal-smectic A-isotropic liquid. The molecular length of the compound is $27.8 \AA$ and the paraffin chain has an all-trans conformation. Intermolecular contacts between two CN groups, between a carbonyloxy and a CN group, and between two carbonyloxy groups are observed in the crystal state.

## Comment

Mesomorphic sequences of 4-cyanophenyl 4-n-alkoxybenzoates ( $\mathrm{CP} n \mathrm{OB}$ ) are dependent on the length of the alkoxy chains, i.e. crystal-nematic-isotropic for $n=5-8$, crystalsmectic A-nematic-isotropic for $n=9-11$, and crystal-smectic A-isotropic for $n=12$ (Vill, 2000). Although the crystal structures of $\mathrm{CP} n \mathrm{OB}$ with shorter alkyl chains $(n=5-8)$ have been reported (Baumeister et al., 1981; Iki \& Hori, 1995), those with longer alkyl chains have not been elucidated. For CP5OB, the crystal has a non-parallel arrangement of molecular long axes (Baumeister et al., 1981). In the crystals of CP6OB and CP8OB (Iki \& Hori, 1995), close contacts between CN groups of a pair of molecules are observed, though in the latter crystal two carbonyloxy groups between another pair of molecules are also closely arranged. For CP7OB (Iki \& Hori, 1995), a CN and a carbonyloxy group come close to each other between a pair of molecules. These compounds are enantiotropic nematogens. We now report the structure of 4-cyanophenyl 4-n-dodecanyloxybenzoate (CP12OB), which has a smectic A phase, with the aim of contributing to a deeper understanding of the relationships between mesomorphic properties and molecular packings in the lattice.

The intersection angle between the least-squares planes $A$ (defined by C2-C7) and $B$ (defined by C9-C14) is 50.5 (1) ${ }^{\circ}$, which is similar to that $\left(49.2^{\circ}\right)$ reported by Iki \& Hori (1995), while that between the least-squares planes $B$ and $C$ (defined by $\mathrm{O} 1, \mathrm{O} 2$ and C 8 ) is $7.9(2)^{\circ}$. The paraffin chain has an alltrans conformation and the molecular length of the compound is 27.848 (13) $\AA$ for the $\mathrm{N} 1 \cdots$ C26 distance.


Intermolecular close contact between CN groups of a pair of antiparallel molecules is observed in the crystal lattice. The distance for $\mathrm{C} 1 \cdots \mathrm{~N} 1^{\mathrm{i}}$ is 3.514 (5) $\AA$ [symmetry code: (i) $-1-x, 1-y, 2-z]$, which is similar to that ( $3.486 \AA$; Iki \& Hori, 1995). Carbonyloxy groups of another pair of anti-


Figure 1
The molecular structure of CP12OB showing $50 \%$ probability displacement ellipsoids.
parallel molecules are also closely arranged. The distances for $\mathrm{O} 1 \cdots \mathrm{O} 1^{\mathrm{ii}}, \mathrm{O} 1 \cdots \mathrm{O} 2^{\mathrm{ii}}$ and $\mathrm{O} 1 \cdots \mathrm{C} 8^{\mathrm{ii}}$ are 3.336 (4), 3.598 (3) and 3.460 (5) $\AA$, respectively [symmetry code: (ii) $-x, 1-y$, $1-z$ ]. In addition, close contacts between a CN and a carbonyloxy group of the other pair of molecules are observed as follows: the distances for $\mathrm{C} 1 \cdots \mathrm{O} 1^{\mathrm{iii}}$ and $\mathrm{N} 1 \cdots \mathrm{O} 1^{\mathrm{iii}}$ are 3.581 (4) and 3.798 (4) $\AA$, respectively [symmetry code: (iii) $-x, 1-y, 2-z]$.

Comparing the dipole moments between a CN group $(4.0 \mathrm{D})$ and a COO group (1.8 D) (The Chemical Society of Japan, 1993), antiparallel interactions between two CN groups is expected to be dominant in the crystal lattice. The crystal of CP12OB, therefore, has a distinct layer structure through infinite networks of the $\mathrm{CN} \cdots \mathrm{CN}$ interaction.

## Experimental

The title compound (CP12OB) was prepared by esterification of 4cyanophenol with 4 -n-dodecanyloxybenzoyl chloride. The single crystals of CP 12 OB were obtained by recrystallization from ethyl acetate.

## Crystal data

$\mathrm{C}_{26} \mathrm{H}_{33} \mathrm{NO}_{3}$
$M_{r}=407.53$
Triclinic, $P \overline{1}$
$a=11.031$ (1) $\AA$
$b=16.639$ (2) $\AA$
$c=6.813$ (1) $\AA$
$\alpha=95.381(10)^{\circ}$
$\beta=94.530(7)^{\circ}$
$\gamma=104.686(10)^{\circ}$
$V=1197.4$ (2) $\AA^{3}$

## Data collection

Enraf-Nonius FR590 diffractometer
$\omega-2 \theta$ scans
Absorption correction: empirical via $\psi$ scans (North et al., 1968)
$T_{\text {min }}=0.961, T_{\text {max }}=0.991$
6224 measured reflections
5747 independent reflections
1764 reflections with $I>2 \sigma(I)$

## Refinement

Refinement on $F^{2}$
$R\left[F^{2}>2 \sigma\left(F^{2}\right)\right]=0.059$
$w R\left(F^{2}\right)=0.193$
$S=0.91$
5747 reflections
272 parameters
H -atom parameters constrained

$$
\begin{aligned}
& Z=2 \\
& D_{x}=1.130 \mathrm{Mg} \mathrm{~m}^{-3} \\
& \text { Mo } K \alpha \text { radiation }
\end{aligned}
$$

Cell parameters from 18
reflections
$\theta=9.1-18.0^{\circ}$
$\mu=0.07 \mathrm{~mm}^{-1}$
$T=296$ (2) K
Prism, colorless
$0.43 \times 0.43 \times 0.10 \mathrm{~mm}$

$$
\begin{aligned}
& R_{\text {int }}=0.039 \\
& \theta_{\max }=28.0^{\circ} \\
& h=-14 \rightarrow 14 \\
& k=-21 \rightarrow 21 \\
& l=-8 \rightarrow 0 \\
& 3 \text { standard reflections } \\
& \quad \text { frequency: } 120 \mathrm{~min} \\
& \text { intensity decay: } 5.7 \%
\end{aligned}
$$

[^0]

Figure 2
Packing diagram viewed down the $a$ axis. H atoms have been omitted for clarity.

Table 1
Selected geometric parameters ( $\left({ }^{\circ},{ }^{\circ}\right)$.

| $\mathrm{N} 1-\mathrm{C} 1$ | $1.133(3)$ | $\mathrm{O} 3-\mathrm{C} 12$ | $1.359(3)$ |
| :--- | ---: | :--- | ---: |
| $\mathrm{O} 1-\mathrm{C} 8$ | $1.188(3)$ | $\mathrm{C} 1-\mathrm{C} 2$ | $1.444(4)$ |
| $\mathrm{O} 2-\mathrm{C} 8$ | $1.371(3)$ | $\mathrm{C} 8-\mathrm{C} 9$ | $1.468(4)$ |
| $\mathrm{O} 2-\mathrm{C} 5$ | $1.400(3)$ |  |  |
| $\mathrm{N} 1-\mathrm{C} 1-\mathrm{C} 2$ | $179.1(4)$ |  |  |
| $\mathrm{O} 2-\mathrm{C} 8-\mathrm{C} 9$ | $110.5(3)$ |  | $172.3(3)$ |
| $\mathrm{C} 8-\mathrm{O} 2-\mathrm{C} 5-\mathrm{C} 4$ | $-48.2(4)$ | $\mathrm{O} 2-\mathrm{C} 8-\mathrm{C} 9-\mathrm{C} 10$ | $179.5(3)$ |
| $\mathrm{C} 5-\mathrm{O} 2-\mathrm{C} 8-\mathrm{O} 1$ | $2.2(5)$ | $\mathrm{C} 12-\mathrm{O} 3-\mathrm{C} 15-\mathrm{C} 16$ |  |
| $\mathrm{C} 5-\mathrm{O} 2-\mathrm{C} 8-\mathrm{C} 9$ | $-176.8(2)$ |  |  |

The positional parameters of the H atoms were calculated geometrically and refined using a riding model. Their $U_{\text {iso }}$ values were fixed to 1.2 times $U_{\text {eq }}$ of the bonded non-H atoms.

Data collection: CAD-4 Software (Enraf-Nonius, 1989); cell refinement: CAD-4 Software; data reduction: MolEN (Fair, 1990); program(s) used to solve structure: SIR97 (Altomare et al., 1999); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: Xtal_GX (Hall \& du Boulay, 1995); software used to prepare material for publication: SHELXL97.

This work was supported by Grant-in-Aid for the Encouragement of Young Scientists (No. 12740381) from the Ministry of Education Science, Sports and Culture, Japan.

## References

Altomare, A., Burla, M. C., Camalli, M., Cascarano, G., Giacovazzo, C., Guagliardi, A., Moliterni, A. G. G., Polidori, G. \& Spagna, R. (1999). J. Appl. Cryst. 32, 115-119.
Baumeister, U., Hartung, H., Graniec, M. \& Jaskólski, M. (1981). Mol. Cryst. Liq. Cryst. 69, 119-130.
Enraf-Nonius (1989). CAD-4 Software. Version 5.0. Enraf-Nonius, Delft, The Netherlands.
Fair, C. K. (1990). MolEN. Enraf-Nonius, Delft, The Netherlands.

Hall, S. R. \& du Boulay, D. (1995). Xtal_GX. University of Western Australia, Australia.
Iki, H. \& Hori, K. (1995). Bull. Chem. Soc. Jpn, 68, 1281-1288.
North, A. C. T., Phillips, D. C. \& Mathews, F. S. (1968). Acta Cryst. A24, 351359.

Sheldrick, G. M. (1997). SHELXL97. University of Göttingen, Germany.
The Chemical Society of Japan (1993). Kagaku Binran Kiso-hen, Vol. II, p. 573. Tokyo: Marzen.

Vill, V. (2000). LiqCryst 3.5. Database of Liquid Crystalline Compounds. Fujitsu Kyushu System Engineering, Fukuoka, Japan.


[^0]:    $w=1 /\left[\sigma^{2}\left(F_{o}{ }^{2}\right)+(0.0774 P)^{2}\right]$
    where $P=\left(F_{o}{ }^{2}+2 F_{c}{ }^{2}\right) / 3$
    $(\Delta / \sigma)_{\max }<0.001$
    $\Delta \rho_{\text {max }}=0.17 \mathrm{e} \AA^{-3}$
    $\Delta \rho_{\text {min }}=-0.17 \mathrm{e}^{-3}$
    Extinction correction: SHELXL97
    Extinction coefficient: 0.0080 (18)

